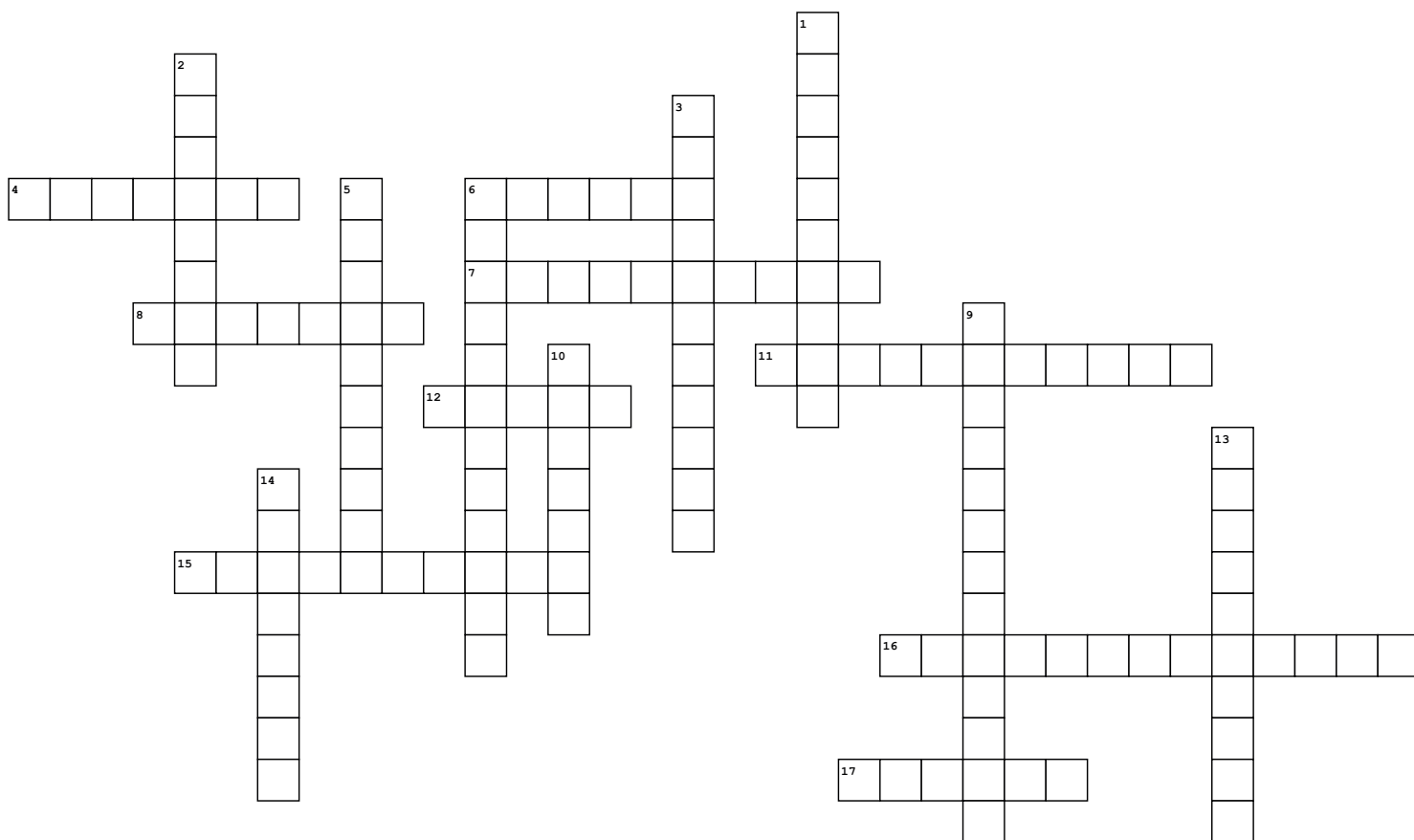


Mixtures



Across

- 4. - Combination of two or more substances
- 6. - Make a solution less concentrated by adding more solvent
- 7. - The maximum amount of solute that can dissolve in a solvent
- 8. - Mixture with particles larger than those in a solution
- 11. - Uniform in composition
- 12. - Homogeneous mixture of metals
- 15. - Heterogeneous mixture with large particles that settle
- 16. - The amount of solute in a given volume of solution
- 17. - Substance being dissolved in a solution

Down

- 1. - Separation technique based on particle size
- 2. - Homogeneous mixture of two or more substances
- 3. - Solid that forms in a solution during a chemical reaction
- 5. - Liquids that do not dissolve in each other
- 6. - Separation technique based on differences in boiling points
- 9. - Not uniform throughout
- 10. - Substance doing the dissolving in a solution
- 13. - Process of dividing a mixture into its components
- 14. - Liquids that can mix in any proportion