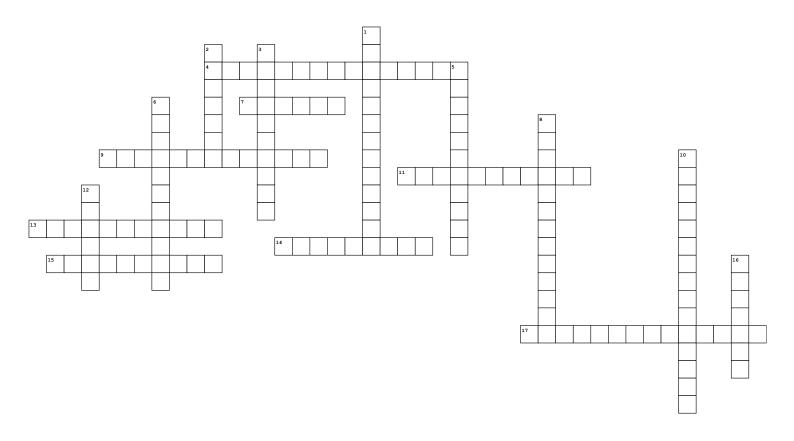
Solutions



Across

- **4.** Pressure exerted by the movement of solvent molecules
- To make a solution less concentrated by adding more solvent
- **9.** The amount of solute in a given volume of solution
- 11. Uniform in composition throughout
- **13.** Properties of a solution dependent on the number of solute particles
- **14.** Process of surrounding solute particles with water molecules
- **15.** Maximum amount of solute that a solvent can dissolve
- **17.** Reaction between an acid and a base to form water and a salt

Down

- 1. Depression in temperature due to solute presence
- 2. Substance capable of dissolving other substances
- 3. Process of forming ions in a solution
- 5. Substance that dissociates into ions in solution
- **6.** Solid formed in a solution during a chemical reaction
- **8.** Pressure exerted by a vapor in equilibrium with its liquid phase
- **10.** Solution containing more solute than it would normally dissolve
- 12. Substance that is dissolved in a solution
- 16. Solution in which water is the solvent