Name:	6. What part of the atom contains the protons and
	neutrons?
Date:	A) The electron cloud
	B) The nucleus
<b>Atoms Science Worksheet</b>	C) The energy levels
TOPIC: ATOMS	D) The outer shell
Instructions: Choose the best answer for each question.	
1. What is the basic unit of matter?	
A) Molecule	7. What is the atomic number of an element?
B) Atom	A) The number of neutrons in the atom
C) Electron	B) The total number of protons and neutrons
D) Proton	C) The number of electrons in the outer shell
,	D) The number of protons in the nucleus
2. What are the three main subatomic particles that	8. How are atoms of different elements different from
make up an atom?	each other?
<ul><li>A) Neutrons, photons, and quarks</li><li>B) Protons, electrons, and neutrons</li></ul>	A) By the number of protons in the nucleus
	B) By the number of neutrons they contain
C) Electrons, ions, and molecules	C) By their size
D) Neutrons, isotopes, and electrons	D) By their energy levels
	b) by their energy levels
3. Where is most of the mass of an atom located?	
A) In the electron cloud	9. What do you call atoms of the same element that have
B) In the nucleus	different numbers of neutrons?
C) In the orbitals	A) Isotopes
D) In the protons	B) Ions
b) in the protons	C) Molecules
	D) Compounds
4. What is the charge of a proton?	
A) Positive	
B) Negative	10. What happens if an atom gains or loses an electron?
C) Neutral	A) It becomes a different element
D) Depends on the element	B) It becomes an ion
	C) It becomes a neutron
	D) It remains unchanged
5. Which subatomic particle has a negative charge?	
A) Neutron	
B) Proton	
C) Electron	
D) Nucleus	
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## Answer Key:

- 1. B
- 2. B
- 3. B
- 4. A
- 5. C
- 6. B
- 7. D
- 8. A
- 9. A
- 10. B